

Supporting information

Novel cage clusters of MoS₂ in the gas phase

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Supporting information 1

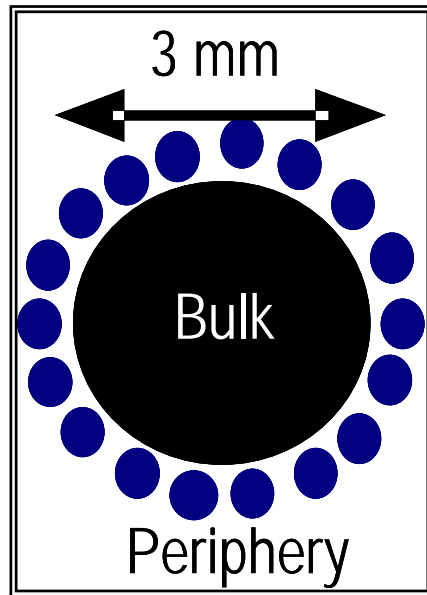


Figure 1.1 Schematic view of the sample spot on the MALDI target plate.

Supporting information 2

PSD mode spectra of some peaks

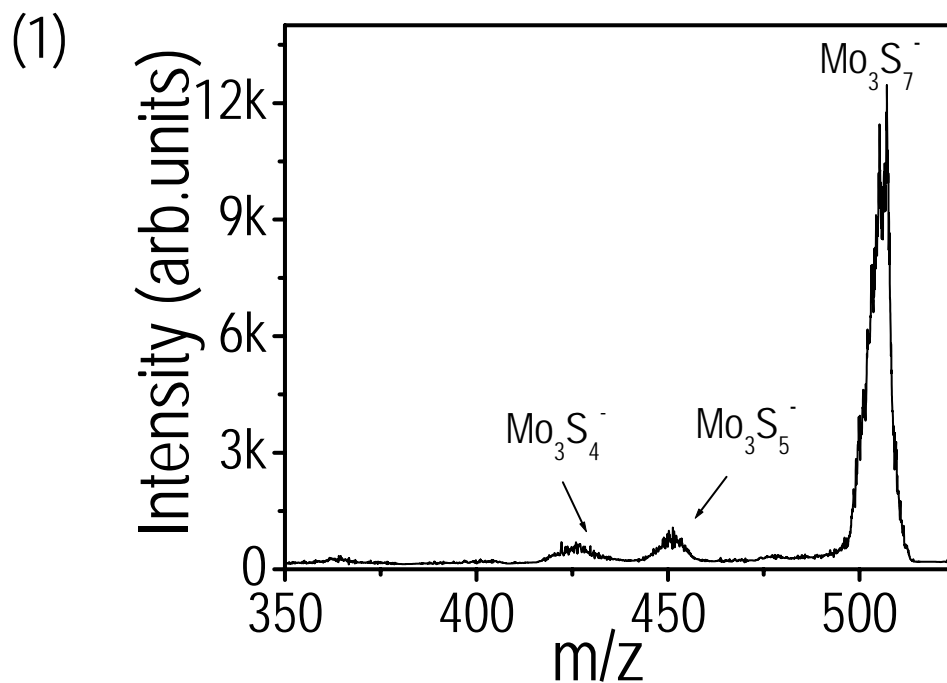


Figure 2.1. PSD spectrum of Mo_3S_7^- ($m/z=513$) showing the fragments at Mo_3S_4^- ($m/z=417$) and Mo_3S_5^- ($m/z=449$).

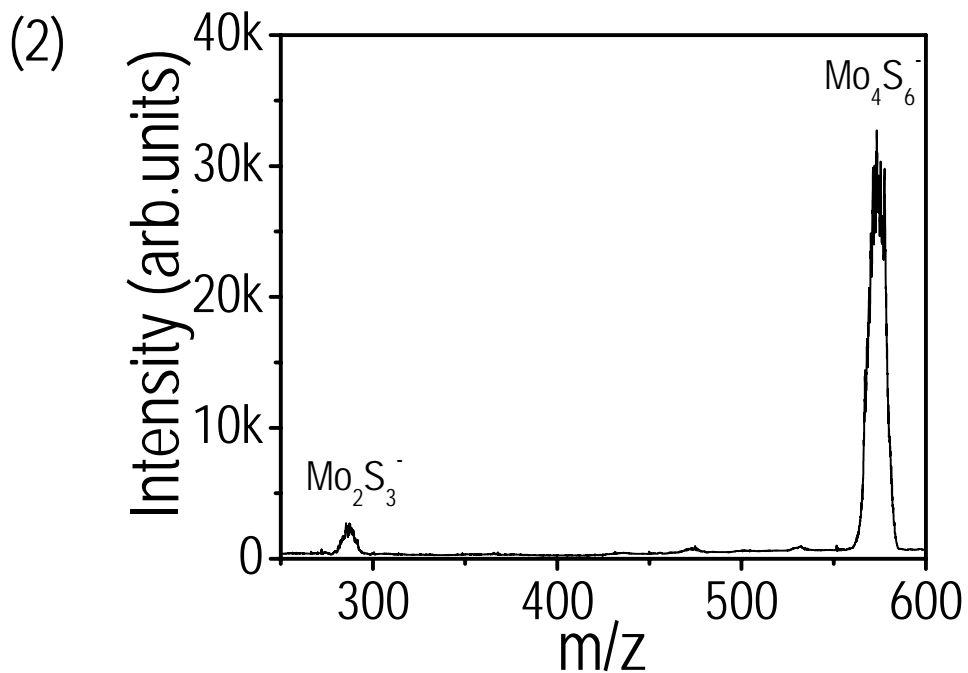


Figure 2.2. PSD spectrum of Mo_4S_6^- ($m/z=574$) showing fragment at Mo_2S_3^- ($m/z=285$).

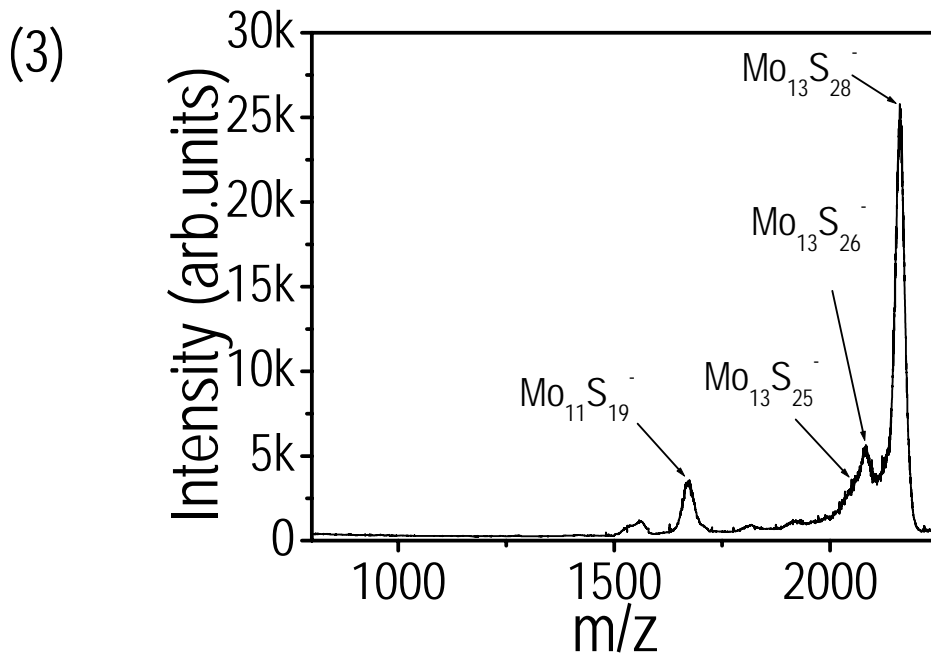


Figure 2.3. PSD spectrum of $\text{Mo}_{13}\text{S}_{28}^-$ showing the fragment ions, $\text{Mo}_{13}\text{S}_{26}^-$ (m/z 2081), $\text{Mo}_{13}\text{S}_{25}^-$ (m/z 2049) and $\text{Mo}_{11}\text{S}_{19}^-$ (m/z 1664).

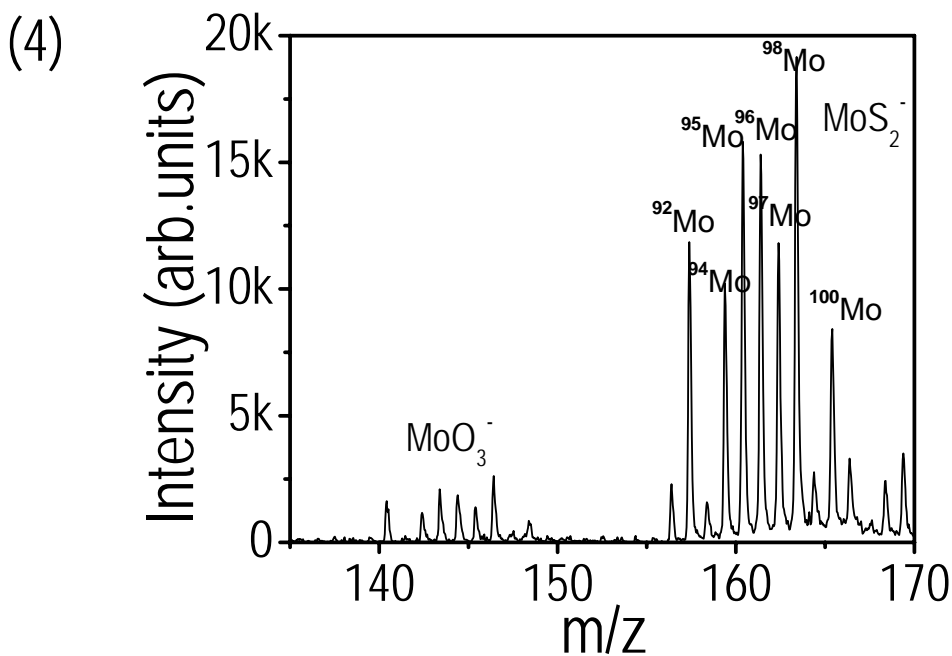


Figure 2.4. Spectrum showing peaks due to MoO_3 and MoS_2 with clear isotope resolution.

Supporting information 3

Various analyses in the extracted MoS₂ flakes.

(1)

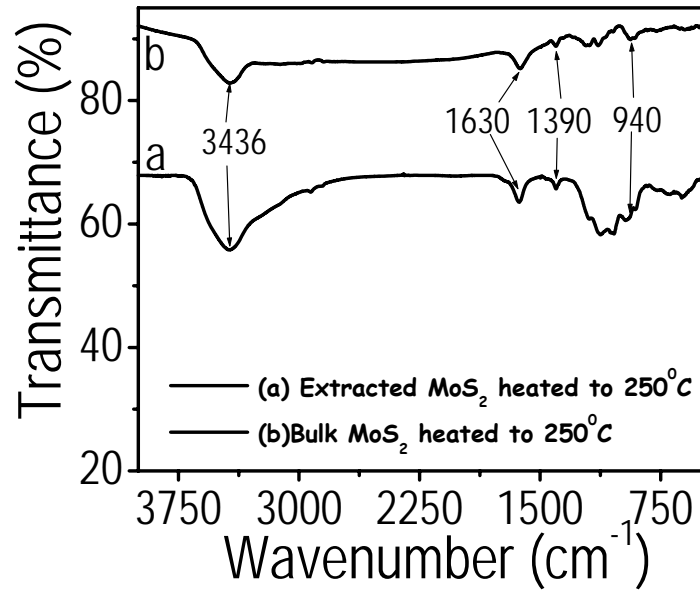


Figure 3.1. The infrared spectrum of the MoS₂ extract (a) is compared with bulk MoS₂ (b). There is a one to one correspondence between the peaks. Minor changes are attributed to the particle size effects (Maugea, F.; Lamotte, J.; Nesterenko, N. S.; Manoilova, O.; Tsyganenko, A. A.; *Catalysis Today* 2001, 70, 271–284).

(2)

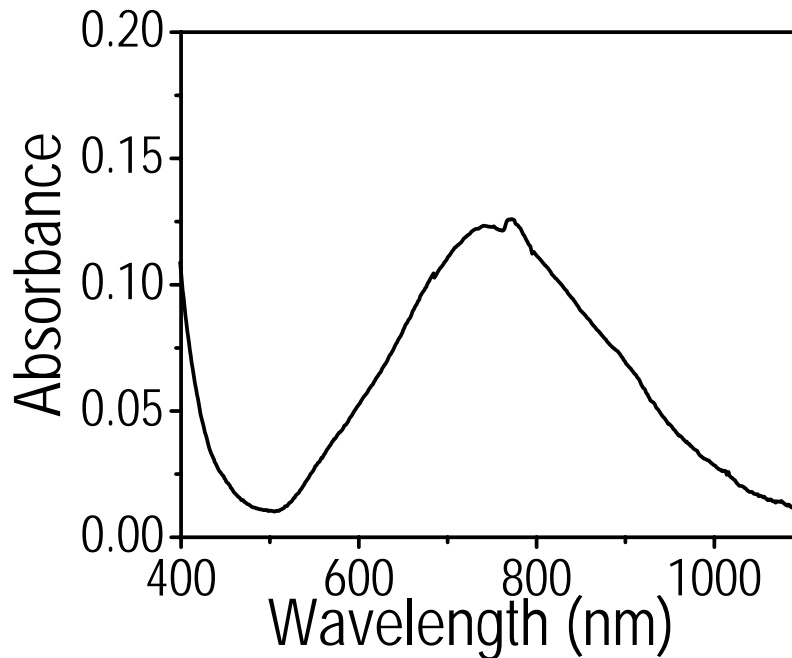


Figure 3.2. UV-Visible spectrum of extracted MoS₂ nano flakes. The peak maximum is at 770 nm.

(3)

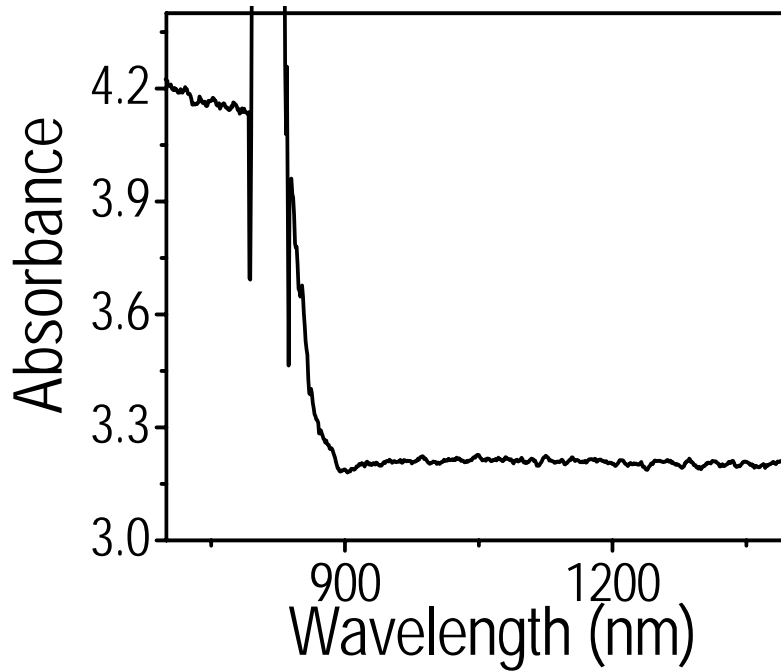


Figure 3.3. UV-Visible diffused reflectance spectrum of bulk MoS₂ showing the bulk band gap of 1.37 eV (Mattheiss, L. F.; *Phys. Rev. Lett.*, 1973, 30, 784-787). The spike around 850 nm is due to a change in the light source of the instrument.

(4)

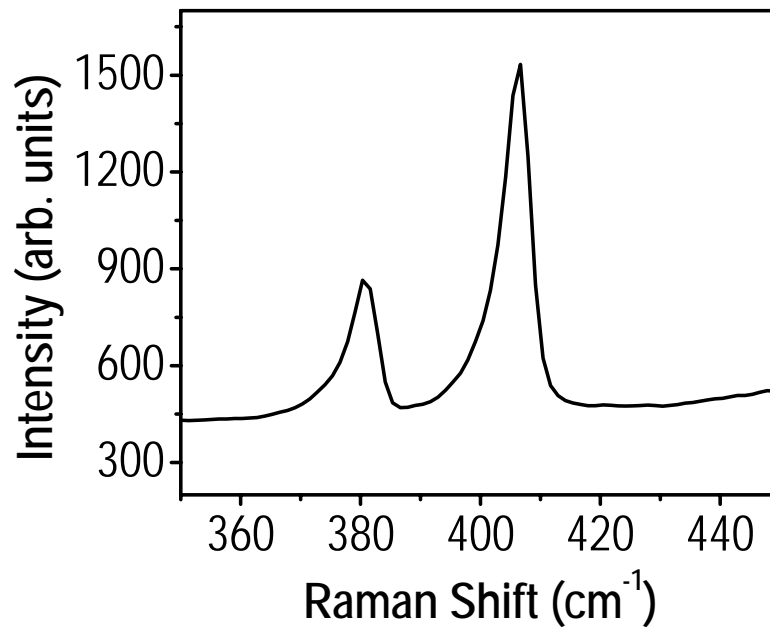


Figure 3.4 Raman spectrum of MoS₂ nanoflakes with 514 nm excitation. The spectrum was measured with a Witec confocal Raman microscope.

Supporting information 4

Spectra in ion selection mode

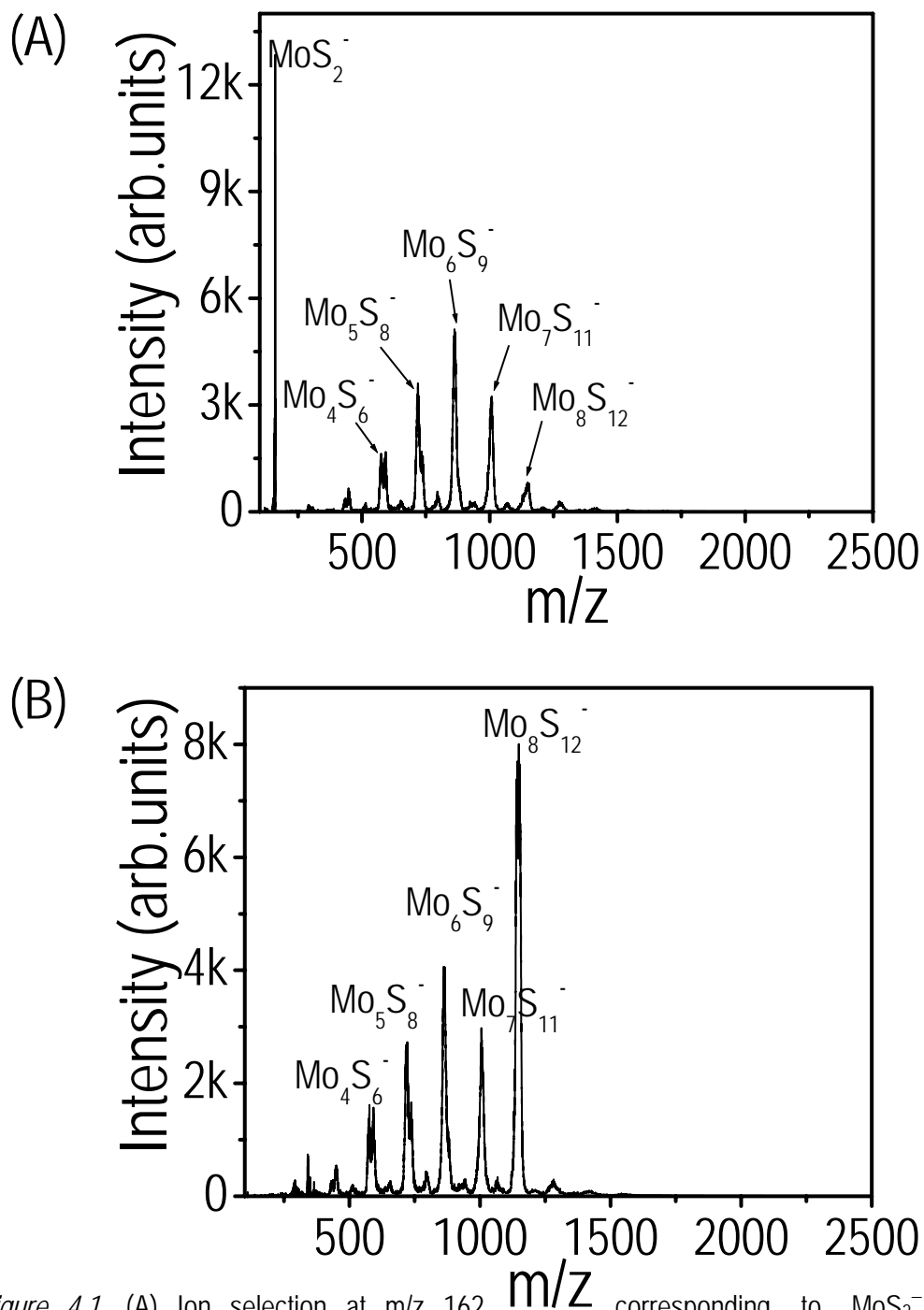


Figure 4.1. (A) Ion selection at m/z 162 corresponding to MoS₂⁻, (B) ion selection at m/z 1146 corresponding to Mo₈S₁₂⁻. While the first one shows extensive clustering, the other shows regular fragmentation. In both Mo₁₃ peak was not present.

Supporting information 5

(1)

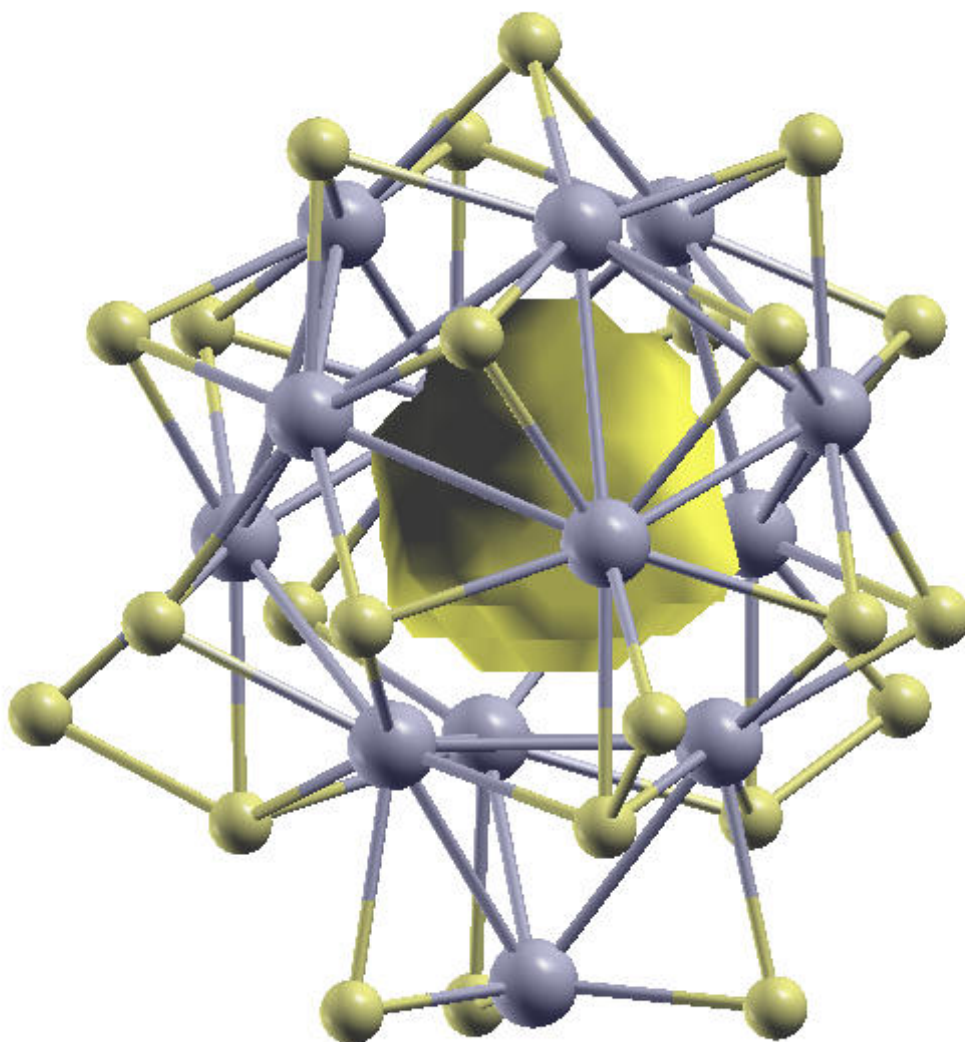


Figure 5.1. Atomic structure of the Mo₁₃S₂₅ cluster. A cloud in the center is plotted as a complement of charge density (region where the density is very close to zero), clearly showing the void space enclosed inside the cage-like structure of the Mo₁₃S₂₅ cluster.

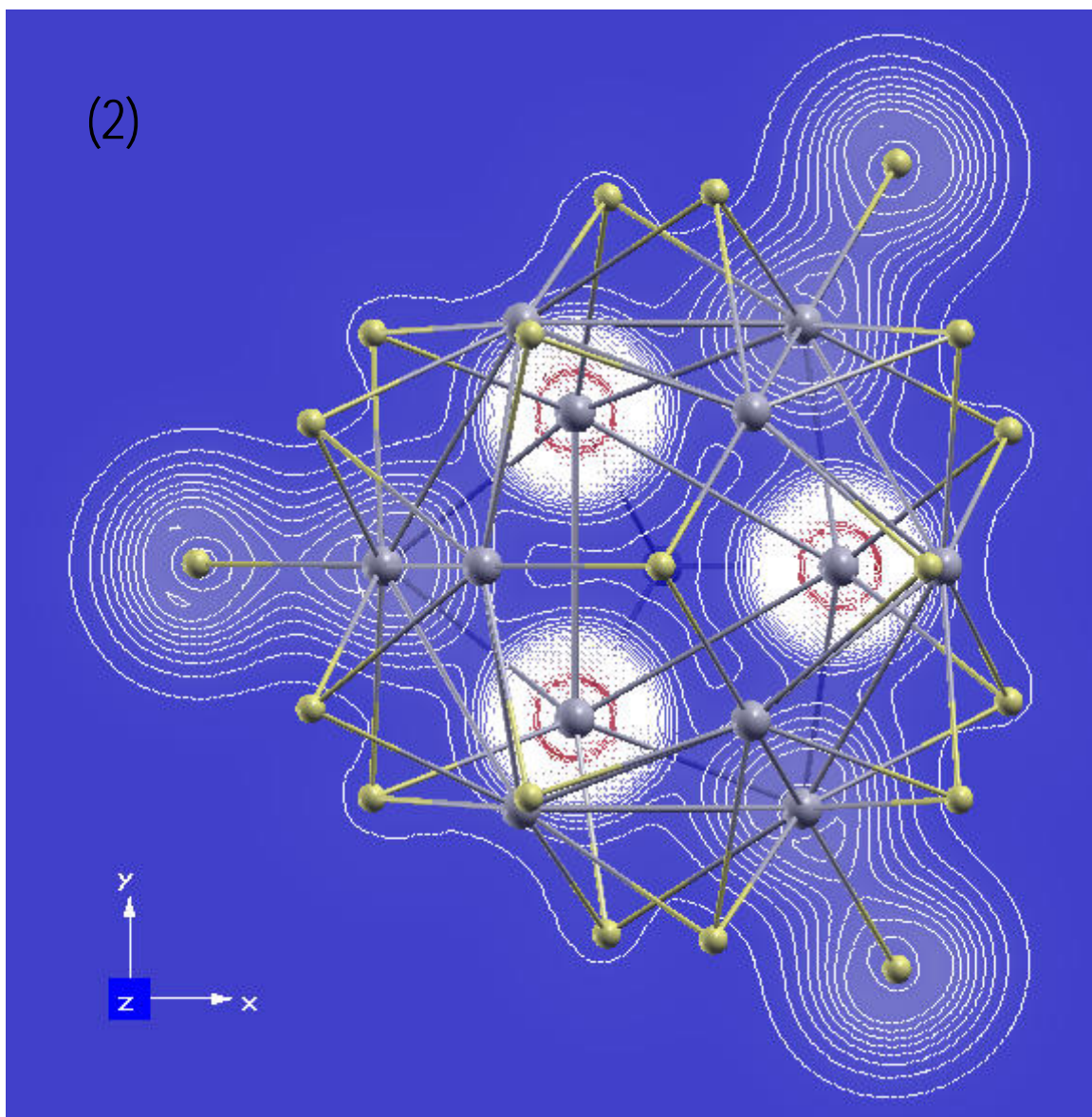


Figure 5.2. Charge density contours of the Mo₁₃S₂₅ cluster, viewed along z-axis (top view of the structure in Figure 1). Plane of the contours is chosen near the bottom triangle of Mo atoms such that it passes through both Mo-S and S-S bonds. Red and White colored rings centered on Mo atoms correspond to fully occupied semi-core 4s and 4p states. Angular contours near the Mo-S bonds indicate their mixed ionic and directional covalent character.

Supporting information 6

WS₂ data

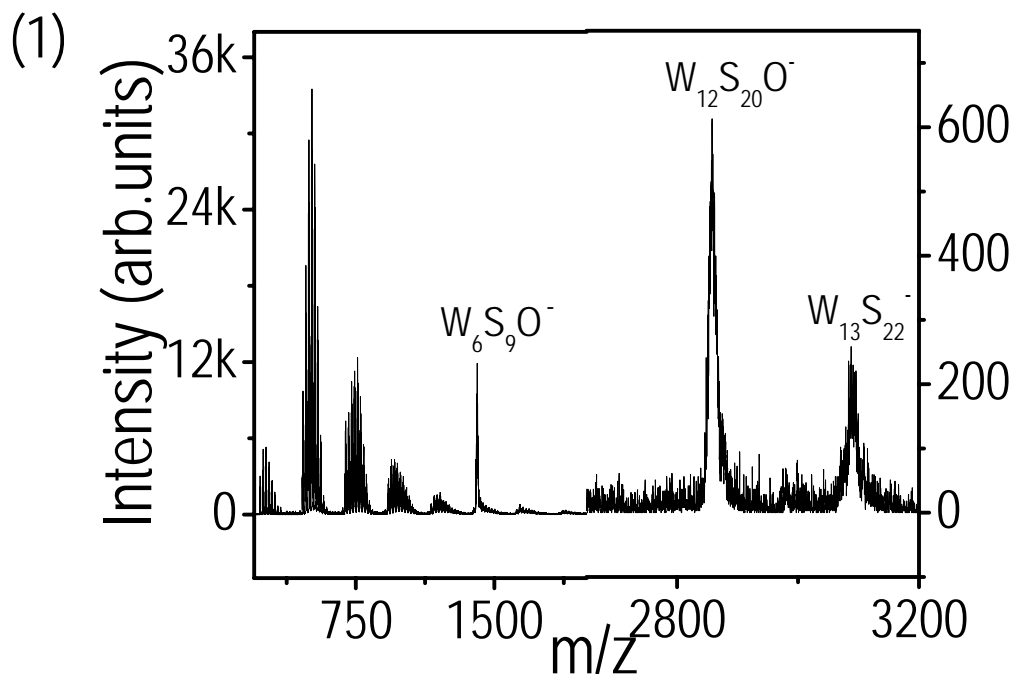


Figure 6.1. LDI spectrum of WS₂ showing magic clusters around the W₆ cluster. The W₁₂ and W₁₃ clusters are zoomed.

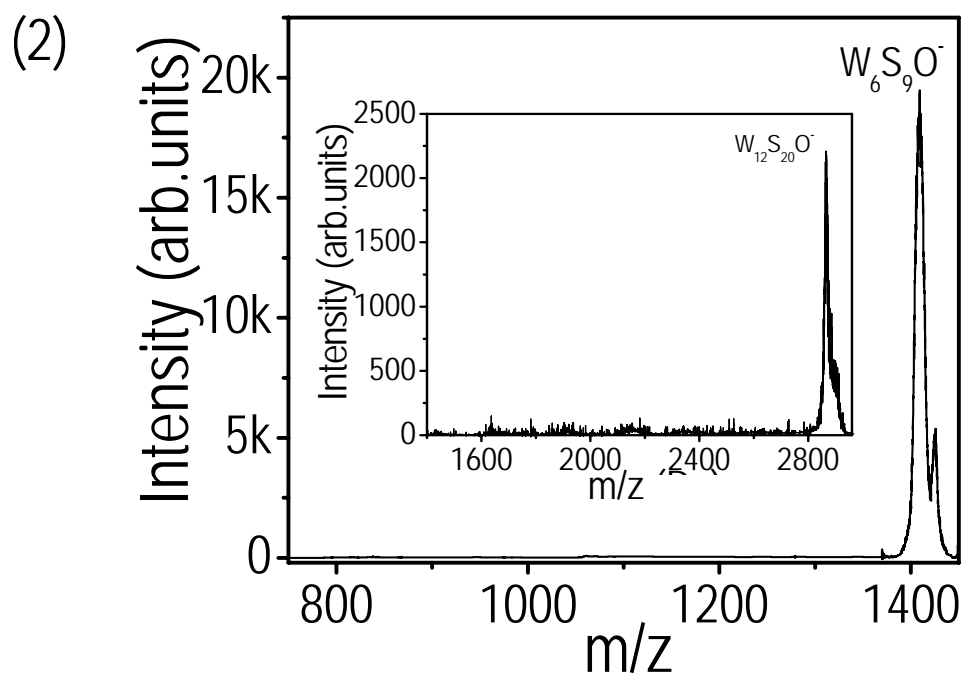


Figure 6.2. PSD mode LDI mass spectrum of W₆S₉O⁻ showing no fragmentation. Inset: PSD of W₁₂S₂₀O⁻ showing no fragmentation.