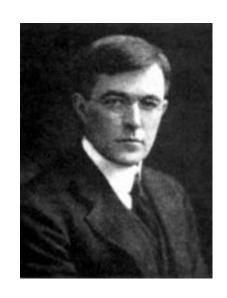
Lecture 3
Surface Dynamics



Irving Langmuir 1881–1957

Concepts of adsorption, desorption, various kinds of adsorption, energetics of adsorption, adsorption isotherms

$$A + S = AS$$
 $\Delta_{ads} H < 0$

Assumptions

- 1. Adsorption does not occur beyond monolayer
- 2. Sites are equivalent and surface is uniform
- 3. Adsorption at one site is independent of occupancy at another site

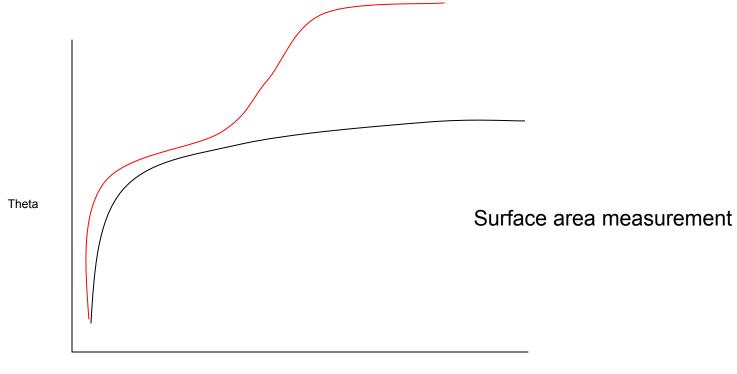
Adsorption,

$$d\Theta/dt = k_a PN(1-\Theta)$$

Desorption,

$$d\Theta/dt = -k_dN\Theta$$

$$P = \frac{\Theta}{K(1 - \Theta)} \quad or \quad \Theta = \frac{KP}{1 + KP}$$



Dissociative Adsorption

$$r\Theta^2 = k(1 - \Theta)^2 P$$

$$\frac{\Theta}{1 - \Theta} = \frac{k^{1/2} P^{1/2}}{r^{1/2}}$$

$$\Theta = \frac{KP^{1/2}}{1 + KP^{1/2}}$$

$$K = (k/r)^{1/2}$$

Adsorption and reaction

